LABDANE DITERPENOIDS FROM NOLANA ROSTRATA

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Abstract—Six new labdane diterpenes, methyl 3-oxo-18-hydroxylabda-8(17),13E-dien-15-oate, methyl 2 β ,3 β -dihydroxylabda-8(17),13E-dien-15-oate, 2-oxo-labda-8(17),13E-dien-15-oate, 3-oxo-18-acetoxylabda-8(17),13E-dien-15-oate, acid, 3-oxo-18-hydroxylabda-8(17),13E-dien-15-oate, acid, 2 β ,3 β -dihydroxylabda-8(17),13E-dien-15-oate, acid, 2 β ,3 β -dihydroxylabda-8(17),13E-dien-15-oate, acid, and the known compound kaempferol-3,7,4'-trimethyl ether were isolated from the aerial parts of Nolana rostrata. The structures of the new compounds were elucidated by spectroscopic methods.

INTRODUCTION

The genus Nolana, distributed in the desert and semidesert zones of Peru and Chile, belong to the Nolanaceae family with ca 18 species arranged into two sections: Alona and Nolana [1]. Very little is known about the secondary metabolites of these species [1]. This paper describes the isolation and structural elucidation of six new labdane derivatives (1a, 1c, 1d, 2a, 2b and 3) present in Nolana rostrata, a typical member of the Alona section, which is characterized by resinous specimens growing in Chile [1].

RESULTS AND DISCUSSION

The petrol extract from the aerial parts of N. rostrata was subjected to column chromatography on silica gel, using increasing proportions of ethyl acetate in petrol to afford kaempferol-3,7,4'-trimethyl ether, methyl 3-oxo-18-hydroxylabda-8(17),13E-dien-15-oate (1a), methyl 2β ,3 β -dihydroxylabda-8(17),13E-dien-15-oate (2a), 2-oxo-labda-8(17),13E-dien-15-oic acid (3), 3-oxo-18-acetoxylabda-8(17),13E-dien-15-oic acid (1c), 3-oxo-18-hydroxylabda-8(17),13E-dien-15-oic acid (1d) and 2β ,3 β -dihydroxylabda-8(17),13E-dien-15-oic acid (2b).

Compound 1a, characterized as its acetate, 1b $(C_{23}H_{34}O_5, [M]^+$ at m/z 390), showed bands for acetoxy, carbonyl and exomethylene groups in the IR spectrum and gave a typical ¹HNMR spectrum for a labda-8(17),13E-diene skeleton (δ 4.94, H-17; 4.62, H-17'; and 2.18, Me-16) with a primary acetoxyl group (δ 4.15, d, J= 11.5 Hz, H-18; 3.98, d, J = 11.5 Hz, H-18', AB system; and 2.05, s, acetyl). The spectrum also showed a threeproton singlet at δ 3.70 corresponding to a carbomethoxyl group, two three-proton singlets at δ 0.99 and 0.88, and a well defined doublet of double doublets at $\delta 2.58$ (J = 16.4, 12.2 and 6.5 Hz) was assigned to an axially orientated proton $(H-2\beta)$ on an adjacent carbon to a ketone function. However, it was the 13 C NMR spectrum of 1b that provided the most information. It confirmed (Table 1) the presence of a ketone function (213.2, s) and an exocyclic methylene group (146.6, s, C-8; 108.0, t, C-17) [2]. The nature and E-geometry of the side chain of 1b

2c R = Ac

Table 1. 13C NMR	spectral data	of compounds	1b, 1c	, 1d,	2a,	2b,	2c,	and	3
	(CDC	1. TMS. SFOR	D)						

Carbon	1b	1c	14	2a	2b	2c	3 54.1	
1	36.0	35.8	36.8	42.8	42.6	40.6		
2	35.3	35.2	35.8	70.8	71.0	69.8	210.4	
3	213.2	213.3	218.0	78.0	78.1	77.9	56.4	
4	50.5	50.4	52.7	38.4	38.6	37.8	37.8	
5	48.0	47.8	48.2	54.4	54.6	54.8	55.0	
6	25.1	24.9	24.7	23.7	23.6	23.6	24.6	
7	37.7	37.6	37.6	38.0	38.1	38.1	37.8	
8	146.6	146.4	147.0	147.0	147.3	146.9	146.8	
9	55.1	55.7	55.9	56.5	57.8	57.7	56.8	
10	39.1	39.1	39.2	38.6	38.4	38.8	37.6	
11	22.1	22.9	22.9	21.7	23.1	22.9	23.2	
12	39.7	32.9	33.0	39.7	33.1	33.1	32.8	
13	160.2	163.4	163.0	160.6	164.1	163.4	163.5	
14	115.3	115.5	115.9	114.8	115.3	115.9	115.6	
15	167.2	171.0	170.8	167.1	170.4	170.4	171.3	
16	18.9	25.8	25.7	18.9	25.8	25.7	25.8	
17	108.0	107.8	107.7	107.1	107.1	108.0	107.9	
18	67.2	67.0	66.5	29.9	29.9	29.5	33.5	
19	17.8	17.8	17.2	17.1	17.1	17.6	22.7	
20	14.0	13.9	14.3	15.7	15.7	15.7	15.4	
COOMe	50.8			50.6		_		
MeCO	170.6	170.5				171.4, 170.6		
MeCO	20.9	20.9				21.2, 20.8	_	

was established by comparison with the 13 C NMR spectral data of methyl copaiferate [3], a labdane-type diterpenoid with identical side chain to that of compound 1b. In fact, the signals due to C-12, C-13, C-14, C-15, C-16 and the methoxyl group of methyl copaiferate appeared at very similar positions to those in the spectrum of 1b (Table 1). The spectrum also established that the CH₂OAc group was equatorially orientated (C-18) at C-4, because a γ -gauche effect on the C-5 was present. In addition, the location of the ketone function at C-3 was again deduced from the 13 C NMR spectrum of 1b which showed the expected deshielding effects on C-2 and C-4 and also γ -effects on the equatorial C-18 oxymethylene group and on the axial Me-19 group, in a similar fashion to that found in methyl 23-hydroxy-3-oxoolean-12-en-28-oate [4].

The assignments of the remaining signals [2, 3] are completely in accord with the proposed structure and 1b thus corresponds to methyl 3-oxo-18-acetoxylabda-8(17),13E-dien-15-oate. The acetate 1b was then hydrolysed with potassium carbonate in methanol and purified by prep. TLC, to give the pure natural product methyl 3-oxo-18-hydroxylabda-8(17),13E-dien-15-oate (1a) (1 H NMR: δ 3.70, d, J = 11.5 Hz, H-18, and 3.36, d, J = 11.5 Hz, H-18'; AB system; IR: 3500 cm⁻¹).

Compound 1c was an amorphous product with molecular formula $C_{22}H_{32}O_5$. Comparison of the ¹H NMR spectrum of 1c with that of 1b showed only minor differences for the skeletal proton signals and differed in the signals due to the side chain. Clearly compound 1c differed from 1b in the geometry of the 13,14 double bond and the ester methyl singlet was missing. As could be deduced from the ¹H NMR spectrum, 1c was a labdane with a Z-configuration double bond $(\delta 1.97, Me-16)$ [5]. The ¹³C NMR spectrum of 1c (Table 1) confirmed this point. The C-12, C-13, C-14, C-15 and C-16 carbon atom

resonances of 1c were almost identical with those of the corresponding carbon atoms reported for labda-8(17)-13Z-dien-15-oic acid [3]. The other carbon resonances remained almost unshifted compared with those of 1b, leading to the assignment of the structure of 1c as 3-oxo-18-acetoxylabda-8(17),13Z-dien-15-oic acid.

The structure of the third compound, 1d ($C_{20}H_{30}O_4$, $[M]^+$ at m/z 334) was readily deduced by comparing its spectral data with that of 1c. In fact, the ¹H NMR spectrum of 1d was very similar to that of 3-oxo-18-acetoxylabda-8(17),13Z-dien-15-oic acid (1c), but the acetate methyl singlet was missing and the H-18 and H-18' signals shifted upfield from δ 4.16 and 3.98 to 3.72 and 3.38, respectively. These differences between the ¹H NMR spectra of 1d and 1c indicated that the new compound (1d) must be the deacetyl derivative of 1c. The ¹³C NMR spectrum of 1d (Table 1) confirmed all the above results and defined the proposed structure as 3-oxo-18-hydroxylabda-8(17),13Z-dien-15-oic acid.

The fourth diterpene, 2a, $C_{21}H_{34}O_4$ ([M] ⁺ at m/z 350) showed absorptions due to hydroxyl and α,β -unsaturated ester groups. The 1HNMR spectrum exhibited the characteristics of a labda-8(17), 13E-diene nucleus (δ 4.90, H-17; 4.64, H-17', and 2.17, Me-16) with an ester methyl signal at δ 3.70, three C-methyl singlets at δ 1.02, 1.00 and 0.96, and two geminal protons to hydroxyl groups [$\delta 4.15$] (1H, q) and 3.23 (1H, d)]. The ¹³C NMR chemical shifts of 2a (Table 1) closely corresponded to those of 1b except for the A-ring carbon atoms, which showed the effects due to two β -oriented hydroxyl groups at C-2 (70.8, d) and C-3 (78.0, d), because significant shifts in the C-1, C-2, C-3, C-4, C-18 and C-19 carbon resonance signals were present as compared to a compound unsubstituted at these carbon atoms (C-2 and C-3) [2]. The assignments of the ¹³C NMR spectral signals of the A-ring carbon atoms of 2a were made on the basis of the observed multiplicities, empirical shift rules $(\alpha, \beta, \text{ and } \gamma)$ effects of the hydroxyl group) and comparison with the reported $^{13}\text{C NMR}$ spectral data of acinosolic acid [6], a triterpenoid with an A ring identical to that of compound 2a. Further confirmation of the β -disposition of the hydroxyl groups was obtained from the $^{1}\text{H NMR}$ spectrum of 2a, since the coupling constant values of its geminal protons indicated that they underwent axial-equatorial coupling. Therefore, on the basis of these data, 2a is shown to be methyl- 2β , 3β -dihydroxylabda-8(17), 13E-dien-15-oate.

Compound 2b, $C_{20}H_{32}O_4$ ([M]⁺ at m/z 336), differs from 2a only in the absence of the ester methyl group and the presence of a Z-configuration for the 13,14 double bond. The ¹H NMR spectrum showed that 2b lacked the ester methyl singlet and the geometry of the side chain was assigned on the basis of the chemical shift of Me-16 (δ 1.92, s), which was as expected for the Z-isomer. The 13CNMR spectrum (Table 1) supports this structural assignment and 2b is shown to be 2β , 3β -dihydroxylabda-8(17),13Z-dien-15-oic acid. Acetylation of 2b afforded two diacetate derivatives, 2c and 2d (see Experimental). The ¹H NMR spectrum of 2c showed the acetate methyl signals at δ 2.03 and the downfield shift of H-2 and H-3 from δ 4.21 and 3.23 in **2b** to 5.38 and 4.65, respectively, in the diacetate 2c. Comparison of the A-ring 13C NMR data obtained for compound 2c with the corresponding data published for acinosolic acid diacetate [6] allowed the location of the substituents as indicated in structure 2c. As could be deduced from the ¹H NMR spectra, 2c and 2d (see Experimental) only differed in the configuration of the side-chain double bond.

The ¹HNMR spectrum of the last compound, 3, $C_{20}H_{30}O_3$ ([M]⁺ at m/z 318) showed the characteristics of a labda-8(17),13Z-diene nucleus (see Experimental) with three C-methyl singlets at $\delta 1.06$, 0.84 and 0.69. However, it was the 13CNMR spectrum of 3 that provided the most information. It showed (Table 1) the presence of a ketone function (210.4, s), and an exomethylene group (146.8, s, C-8; 107.9, t, C-17). The nature and Z-geometry of the side chain of 3 was unambiguously established by comparison of its 13C NMR spectral data with those of the corresponding carbon atoms of 1c, 2b and 2c. The location of the ketone function at C-2 was also deduced from the ¹³C NMR spectrum, which showed the expected deshielding effects on C-1 and C-3. These assignments are in agreement with those reported for 2keto-manoyl oxido [3]. Therefore, compound 3 corresponded to 2-oxo-labda-8(17),13Z-dien-15-oic acid.

EXPERIMENTAL

Mps uncorr. ¹H NMR: 60, 100 or 360 MHz in CDCl₃ with TMS. ¹³C NMR: 22.15 or 90.5 MHz. IR: film, CHCl₃ or KBr pellets. MS: direct inlet, 70 eV.

Nolana rostrata (Lindley) Mier, was collected in La Serena, IV-Región, Chile, in Sept. 1984. A voucher specimen is deposited at Universidad Federico Santa María.

The air dried aerial parts (1.0 kg) of N. rostrata were extracted at room temp. with petrol for 24 hr, affording 15.8 g of a syrup. A portion of this syrup (10 g) was chromatographed on a silica gel column (400 g) eluting with mixtures of petrol and EtOAc of increasing polarity. Fractions of 100 ml were taken and combined based upon TLC monitoring. Fraction 10 provided kaempferol-3,7,4'-trimethyl ether (110 mg). Fraction 33 gave a mixture containing 1a (280 mg). Fractions 37-42 provided a mixture of 2a

and 3. Fractions 77-79, after crystallization from *n*-heptane-EtOAc, gave 1c (140 mg). Fractions 92 and 113-115 gave 1d (200 mg) and 2b (151 mg), respectively.

Purification of fraction 33. The ¹H NMR spectrum of fraction 33 suggested the presence of one major compound with a primary OH group (1a). This fraction, after treatment with pyridine (0.5 ml) and Ac₂O (2.0 ml) at room temp. overnight and evapn of the reagents under vacuum, was chromatographed on a silica gel column (20 g) and eluted with petrol-EtOAc (8:2), yielding 1b (230 mg).

Purification of 2a and 3. Fractions 37-42 (250 mg) were rechromatographed on a silica gel column (30 g) and eluted with petrol-EtOAc (3:2) yielding pure 2a (110 mg) and 3 (40 mg).

Methyl 3-oxo-18-hydroxylabda-8(17),13E-dien-15-oate (1a). Compound 1b (50 mg) was hydrolysed with K_2CO_3 in MeOH at room temp. under N_2 . After 1 hr the mixture was filtered, concd and purified by prep. TLC (R_f 0.55, silica gel, petrol-EtOAc, 3:2) to give 1a (28 mg), gummy. IR $v_{max}^{CHCl_3}$ cm⁻¹: 3500, 3080, 3020, 2960, 2840, 1710, 1700, 1640, 1450, 1440, 1380, 1160, 900. ¹H NMR (60 MHz): δ5.65 (1H, br s, H-14), 4.93 (1H, br s, H-17), 4.58 (1H, br s, H-17'), 3.67 (3H, s, COOMe), 3.70 (1H, d, J = 11.5 Hz, H-18), 3.36 (1H, d, J = 11.5 Hz, H-18'), 2.60-2.30 (4H, m, H-2β, H-7α, H-9α and H-2α), 2.17 (3H, br s, H-16), 0.93 (3H, s, H-19), 0.90 (3 H, s, H-20). MS m/z (rel. int.): 348 [$C_{21}H_{32}O_4$, M] + (25), 333 [M - Me] + (17), 330 [M - H₂O] (38), 297 (24), 235 (15), 217 (40), 114 (38), 91 (59), 82 (100), 55 (70).

3-Oxo-18-acetoxylabda-8(17),13Z-dien-15-oic acid (1c). Amorphous powder (n-hexane-EtOAc); [a] β + 37.5° (CHCl₃; c. 1.10). IR $\nu_{\text{max}}^{\text{KBr}}$ cm $^{-1}$: 3500, 3080, 3020, 2960-2840, 1740, 1700, 1640, 1450, 1415, 1380, 1250, 1050, 900. 1 H NMR (100 MHz): δ 5.68 (1H, br s, H-14), 4.95 (1H, br s, H-17), 4.77 (1H, br s, H-17'), 4.16 (1H, d, J = 11.4 Hz, H-18), 3.98 (1H, d, J = 11.4 Hz, H-18') 2.60-2.30 (4H, m, H-2 β , H-7 α , H-9 α , H-2 α), 2.07 (3H, s, CH₂OAc), 1.97 (3H, br s, H-16), 0.99 (3H, s, H-19), 0.87 (3H, s, H-20). 13 C NMR: see Table 1. MS m/z (rel. int.): 376 [C₂₂H₃₂O₅, M]⁺ (2), 358 [M-H₂O]⁺ (2), 316 [M-HOAc]⁺ (4), 277 [M-C₅H₆O₂]⁺ (9), 217 [277 - HOAc] (7), 91 (40), 81 (40), 79 (42), 67 (49), 55 (70), 43 (100), 41 (90).

3-Oxo-18-hydroxylabda-8(17),13Z-dien-15-oic acid (1d). Mp 84-86° (n-heptane-EtOAc). [α] $^{25}_{D}$ +28.5° (CHCl $_3$; c 2.00). IR v_{max}^{KBr} cm $^{-1}$: 3550, 3400, 3080, 2960, 2840, 1700, 1640, 1450, 1415, 1260, 1050, 940, 900, 860. 1 H NMR (100 MHz): δ 5.68 (1H, br s, H-14), 4.95 (1H, br s, H-17), 4.76 (1H, br s, H-17), 3.72 (1H, d, J = 11.4 Hz, H-18), 3.38 (1H, d, J = 11.4 Hz, H-18'), 2.60-2.30 (4H, m, H-2 β , H-7 α , H-9 α , H-2 α), 1.96 (3H, s, H-16), 0.95 (3H, s, H-19), 0.90 (3H, s, H-20). 13 C NMR: see Table 1. MS m/z (rel. int.): 334 [C₂₀H₃₀O₄, M] $^+$ (3), 316 [M - H₂O] $^+$ (12), 304 (27), 287 (11), 235 [M - C₅H₇O₂] $^+$ (36), 206 (25), 133 (40), 116 (46), 107 (53), 95 (52), 81 (51), 79 (52), 67 (39), 55 (69), 43 (100), 41 (77).

Methyl-2 β ,3 β -dihydroxylabda-8(17),13E-dien-15-oate (2a). Oil. [α] $_{0}^{25}$ – 52.8° (CHCl₃; c 1.08). IR $v_{\rm max}^{\rm nim}$ cm $^{-1}$: 3500, 3080, 3020, 2960–2840, 1715, 1710, 1640, 1450, 1230, 1150, 1040, 900,

760. 1 H NMR (100 MHz): δ 5.65 (1H, q, J = 1 Hz, H-14), 4.90 (1H, br s, H-17), 4.64 (1H, br s, H-17'), 4.15 (1H, q, J = 4.0 Hz, H-2 α), 3.70 (3H, s, COOMe), 3.23 (1H, d, J = 4.0 Hz, H-3 α), 2.50-2.30 (2H, m, H-7 α and H-9 α), 2.17 (3H, d, J = 1.0 Hz, H-16), 1.02 (3H, s, H-18), 1.00 (3H, s, H-19), 0.96 (3H, s, H-20). 13 C NMR: see Table 1. MS m/z (rel. int.): 350 [C₂₁H₃₄O₄, M]⁺ (35), 335 [M - Me]⁺ (12), 332 [M - H₂O]⁺ (24), 317 [M - H₂O - Me]⁺ (21), 133 (13), 121 (21), 109 (45), 91 (27), 81 (35), 55 (36), 43 (100), 41 (60).

 2β , 3β -Dihydroxylabda-8(17), 13Z-dien-15-oic acid (2b). Amorphous powder (n-heptane-EtOAc). [α] $_{25}^{55}$ - 16.7° (CHCl₃; c 0.61). IR ν $_{max}^{KBr}$ cm $^{-1}$: 3500, 3080, 2960, 2840, 1700, 1650, 1450, 1380, 1250, 1070, 1040, 900, 760. 1 H NMR (100 MHz): δ 5.64 (1H, br s, H-14), 4.88 (1H, br s, H-17), 4.62 (1H, br s, H-17), 4.21 (1H, q, J = 4.0 Hz, H-2 α), 3.23 (1H, d, J = 4.0 Hz, H-3 α), 2.45-2.25 (2H, m, H-7 α and H-9 α), 1.92 (3H, s, H-16), 0.98 (3H, s, H-18), 0.95 (3H, s, H-19), 0.91 (3H, s,H-20). 13 C NMR: see Table 1. MS m/z (rel. int.): 336 [C₂₀H₃₂O₄, M] $^+$ (3), 318 [M - H₂O] $^+$ (3), 303 [318 - Me] $^+$ (3), 300 [M - 2H₂O] $^+$ (4), 285 (6), 121 (100), 105 (80), 95 (73), 85 (66), 79 (97), 67 (54), 55 (80), 43 (76).

2β,3β-Diacetoxylabda-8(17),13Z-dien-15-oic acid (2c) and 2β , 3β -diacetoxylabda-8(17), 13E-dien-15-oic acid Compound 2b (100 mg) was treated with Ac₂O (3.0 ml) and pyridine (0.5 ml) at room temp, for 36 hr. After addition of EtOH, the reaction mixture was evapd to dryness under vacuum. This crude material showed two components by TLC. The substances were isolated by prep. TLC (silica gel; petrol-EtOAc, 7:3) and yield 2c (50 mg R_c 0.60) and 2d (20 mg, R_c 0.55). 2c, white crystals, mp 122–125° (n-heptane–EtOAc). $[\alpha]^{25} - 55.5^{\circ}$ $(CHCl_3; c 1.0)$. IR v_{max}^{KBr} cm⁻¹: 3450, 3080, 3020, 2960, 2840, 1740, 1700, 1650, 1450, 1380, 1250, 1040, 960, 890. ¹H NMR (60 MHz): δ 5.70 (1H, br s, H-14), 5.38 (1H, q, J = 4.0 Hz, H-2 α), 4.94 (1H, br s, H-17), 4.75 (1H, br s, H-17'), 4.65 (1H, d, J = 4.0 Hz, H-3 α), 2.03 (6H, s, 2 × OAc), 1.93 (3H, s, H-16), 1.03 (3H, s, H-18), 0.90 (6H, s, H-19 and H-20). 13 C NMR: see Table 1. MS m/z (rel. int.): 420 $[C_{24}H_{36}O_6, M]^+$ (15), 402 $[M-H_2O]^+$ (15), 360 [M] $-HOAc]^+$ (40), 318 [360-CH₂=C=O]⁺ (48), 300 [318 $-H_2O]^+$ (97), 285 (98), 261 (90), 219 (83), 201 (99), 131 (84), 43 (100). Compound **2d**, gummy. $[\alpha]_D^{25} - 30.3^{\circ}$ (CHCl₃; c 0.60). IR and MS identical to those of 2c. ¹H NMR (60 MHz): δ5.70 (1H,

br s, H-14), 5.40 (1H, q, J = 4.0 Hz, H-2 α), 4.97 (1H, br s, H-17), 4.67 (1H, d, J = 4.0 Hz, H-3 α), 4.58 (1H, br s, H-17'), 2.16 (3H, s, H-16), 2.04 (6H, s, 2 × OAc), 1.02 (3H, s, H-18), 0.90 (6H, s, H-19 and H-20).

2-Oxo-labda-8(17),13Z-dien-15-oic acid (3). Gum. [α] $\frac{2}{15}$ + 55.4° (CHCl₃, c 0.81). IR $v_{\rm max}^{\rm him}$ cm $^{-1}$: 3400, 3080, 2960–2840, 1700, 1650, 1450, 1415, 1380, 1250, 900. 1 H NMR (100 MHz): δ5.68 (1H, br s, H-14), 4.94 (1H, br s, H-17), 4.73 (1H, br s, H-17), 2.60–2.50 (2H, m, H-1α and H-3α), 1.93 (3H, br s, H-16), 1.06 (3H, s, H-18), 0.84 (3H, s, H-20), 0.69 (3H, s, H-19). 13 C NMR: see Table 1. MS m/z (rel. int.): 318 [C₂₀H₃₀O₃, M] $^{+}$ (34), 303 [M $^{-}$ Me] $^{+}$ (25), 300 [M $^{-}$ H₂O] $^{+}$ (32), 285 [M $^{-}$ Me $^{-}$ H₂O] $^{+}$ (22), 219 [M $^{-}$ C₅H₇O₂] $^{+}$ (57), 151 (65), 135 (30), 91 (68), 82 (90), 69 (70), 55 (82), 43 (62), 41 (100).

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